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The electron configuration for an element  
can be represented using the noble-gas  
notation in the previous \_\_\_\_\_ and the

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Electron configuration for the energy level being filled. period. The electrons in an atom's outermost orbitals are called \_\_\_\_\_. Valence electrons. In an electron-dot structure, the element's symbol represents the \_\_\_\_\_. ...

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(Electron configuration) ?questionThe

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Actual Electron Configurations. •Total electrons = atomic number •Fill energy levels with electrons until you run out •A superscript states how many electrons are in each level. –Hydrogen – $1s^1$ –1 electron

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Answers – Helium –  $1s^2$  – 2 electrons total.

– Lithium –  $1s^2 2s^1$  – 3 electrons total.

## Electron Configuration (Section 5.2)

38. The ground state electron configuration of Fe is \_\_\_\_\_. A.  $2 1s 2s 2 3s 3p^6 3d^6$  B.  $2 1s 2 2s 2p^6 3s^2 3p^6 4s 3d^6$  C.  $2 1s^2 2s 2p^6 3s^2 3p^6 4s^2$  D.  $2 1s 2$

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2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup> 4s<sup>2</sup> 4d<sup>6</sup> 39. The ground state electron configuration of Ga is \_\_\_\_\_.  
A. 2 1s 2s<sup>2</sup> 3s<sup>2</sup> 3p<sup>6</sup> 3d<sup>10</sup> 4s<sup>2</sup> 4p<sup>1</sup>  
B. 6 1s 2 2s 2p 3s<sup>2</sup> 3p<sup>6</sup> 4s<sup>2</sup> 4d<sup>10</sup> 4p<sup>1</sup> C. 2 1s<sup>2</sup> 2s 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup> 4s<sup>2</sup> 3d<sup>10</sup> 4p<sup>1</sup>

Atomic Structure and Electron Configurations Multiple ...



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**Answer:** All have an  $ns^2 np^5$  electron configuration, one electron short of a noble gas electron configuration. (Note that the heavier halogens also have filled  $(n-1)d^{10}$  subshells, as well as an  $(n-2)f^{14}$  subshell for Rn; these do not, however, affect their chemistry in any significant way.

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6 Answers and Worksheet Electron  
Configuration Pdf Kidz Activities. The  
Key Answers for the Electron  
Configuration Chem Worksheet key

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Answer, together with the other answers give the student, after doing the associated experiments, the knowledge to make a reasonable judgment as to what to do to minimize and eliminate the possibility of such a reaction in ...

Electron Configuration Chem Worksheet 5

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1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup>. The number of valence electrons in the atom is The number of valence electrons in the atom is answer choices

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The homework is practice of determining subatomic particles and beginning electron configuration. The homework is due the

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Answers where I stamp it for completion and go over answers using the answer key. Most students do well on the homework with the only common mistake of placement of electrons in energy shells (2, 8, 18)

Ninth grade Lesson Electron

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## Configuration Part I ...

$n$  determines the general range for the value of energy and the probable distances that the electron can be from the nucleus.  $l$  determines the shape of the orbital.  $m_l$  determines the orientation of the orbitals of the same  $l$  value with respect to one another.  $m_s$  determines the spin of an

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